

# Monitoring radiation levels in the Pacific Northwest

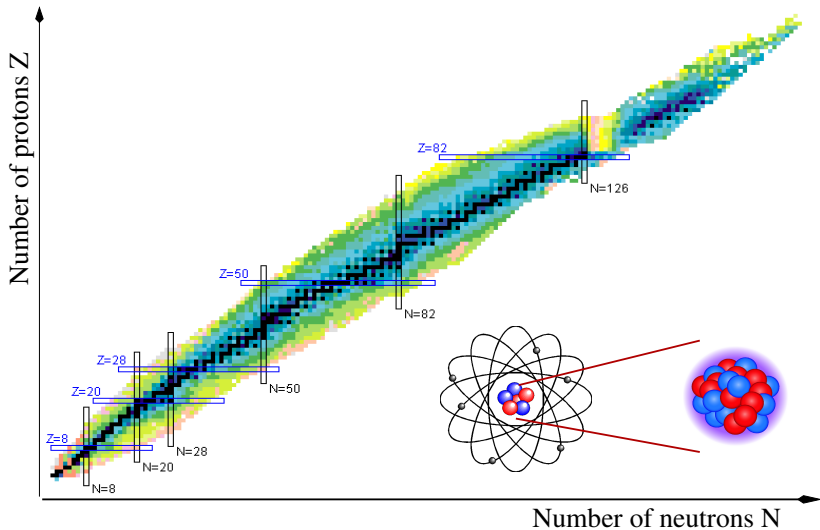
## The Fukushima Nuclear Crisis: Separating Fact from Fiction

Simon Fraser University

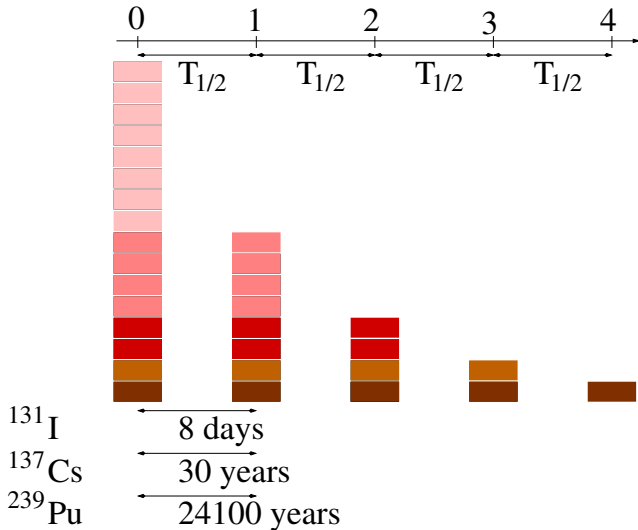
April 11, 2011



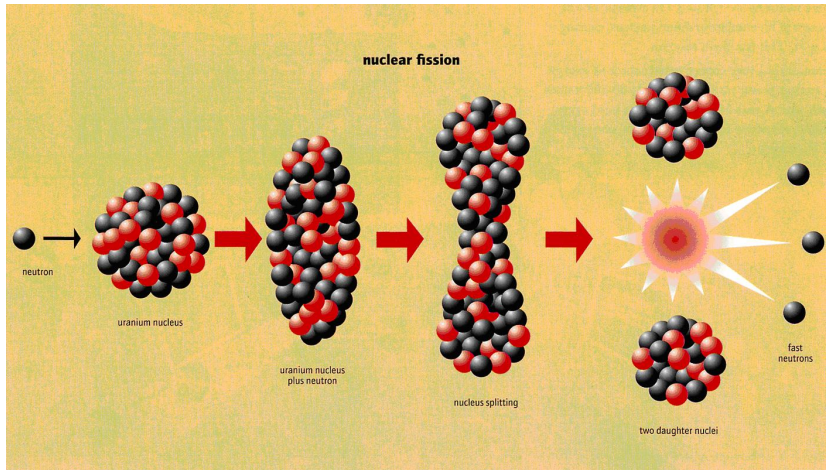
# The nuclear chart



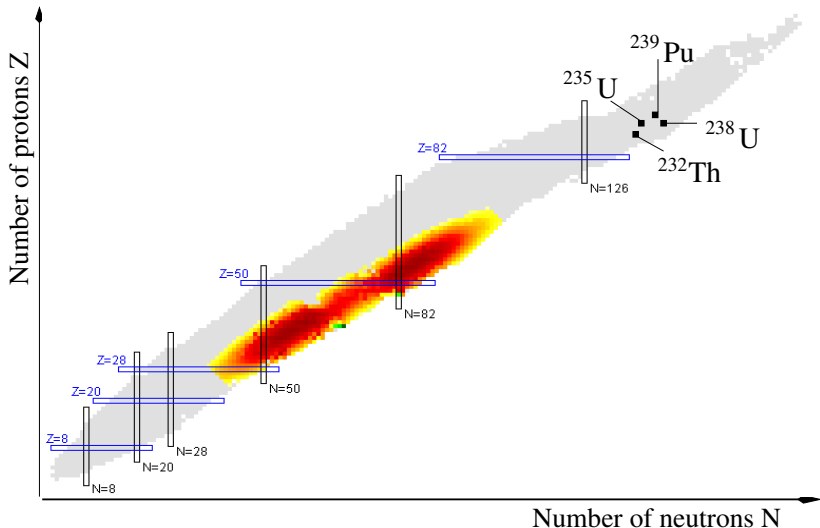
# Nuclear decay half-life



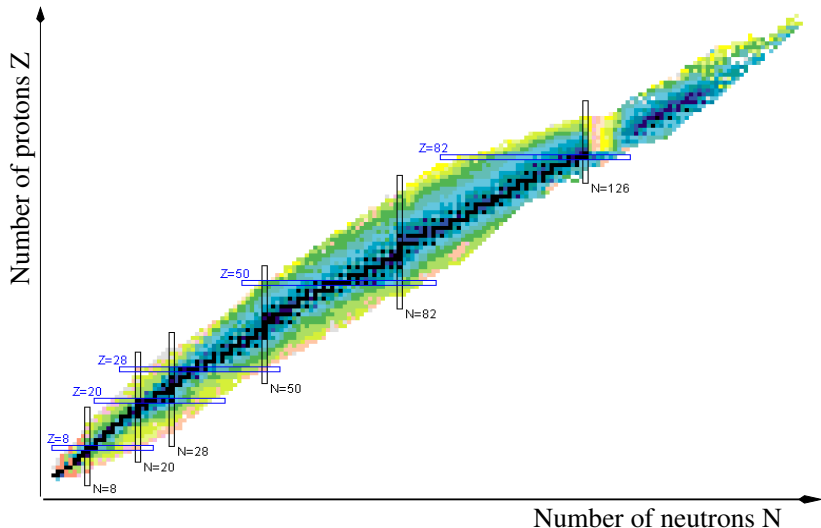
# Nuclear fission



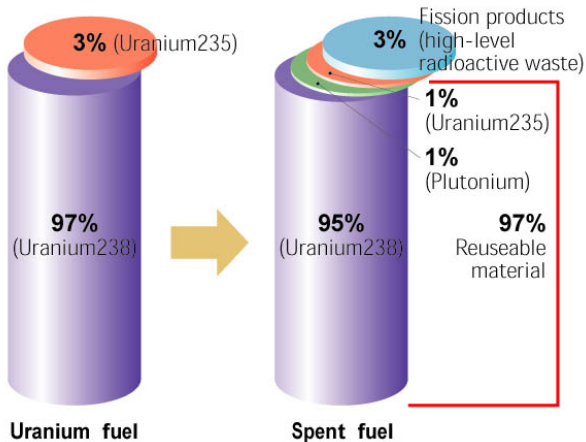
# Fission fragments



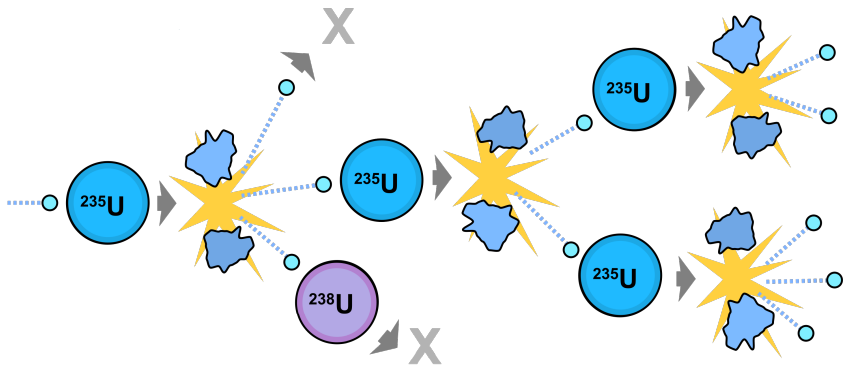
# The nuclear chart



# Fresh and spent nuclear fuel



# Nuclear chain reaction





# Plutonium

- Plutonium (Pu) is a man made, radioactive, heavy element containing 94 protons.
- $^{239}\text{Pu}$  is produced in fuel containing  $^{238}\text{U}$ .
- $^{239}\text{Pu}$  is the material of choice for nuclear weapons.
- Mixed Oxide nuclear fuel used in one of the Fukushima reactors contains Pu/U mixture.
- The element of Pu is highly toxic when inhaled, (less toxic when ingested).
- Animal studies found that an accumulated dose of a few milligram of plutonium per kilogram of tissue is lethal.
- Traces of Pu were reported to be found at the Fukushima site (but not talked about recently).

# Radio-iodine $^{131}\text{I}$

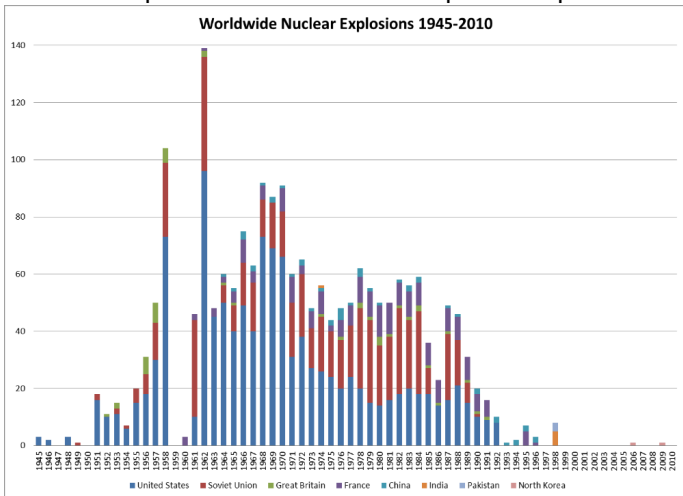
- $^{131}\text{I}$  is a man made radioactive isotope of Iodine.
- $^{131}\text{I}$  has a half life of 8 days.
- $^{131}\text{I}$  is an abundant fission fragment.
- $^{131}\text{I}$  is used in radiotherapy of cancer.
- $^{131}\text{I}$  is not found in the atmosphere in normal conditions .
- $^{131}\text{I}$  is a good indicator of radioactivity release in reactor accidents, the signal is not obstructed by background.
- Iodine accumulates in the thyroid, thus large scale exposure to  $^{131}\text{I}$  is a potential health hazard.

# Radioactive $^{137}\text{Cs}$

- $^{137}\text{Cs}$  is a man made radioactive isotope of Cesium.
- $^{137}\text{Cs}$  has a half life of 30 years.
- $^{137}\text{Cs}$  is an abundant fission fragment.
- $^{137}\text{Cs}$  is used in radiotherapy of cancer and in food irradiation.
- $^{137}\text{Cs}$  was not present in the environment before 1940's when fission started to be used for nuclear power releases.
- $^{137}\text{Cs}$  background from nuclear weapon tests and previous nuclear accidents obscures the signal from Fukushima.
- Contamination with  $^{137}\text{Cs}$  is long lasting.

# Nuclear weapon tests

~2000 reported tests ~550 atmospheric explosions

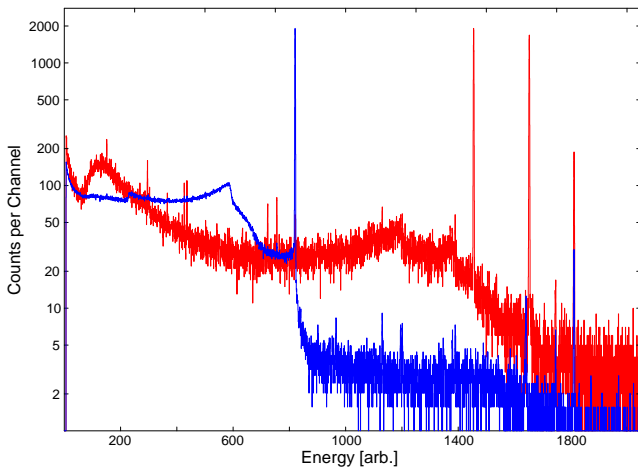


# Fission fragments release

- Release depends critically on the accident scenario.
- Fission fragments which exists as gases, vapours, or aerosols are released first.
- The Three Mile Island accident released gases only, including  $^{131}\text{I}$ .
- The explosions and fires following the Chernobyl accident resulted in a release of 6 tons of fragmented fuel along with radioactive gases.
- The release of  $^{131}\text{I}$  from the Chernobyl was 2.4 million times larger than from the Three Mile Island accident.
- The crisis management at Fukushima successfully prevented large-scale radioactivity release in the past month, however, the cooling operates under emergency conditions.

# Characteristic decay spectra

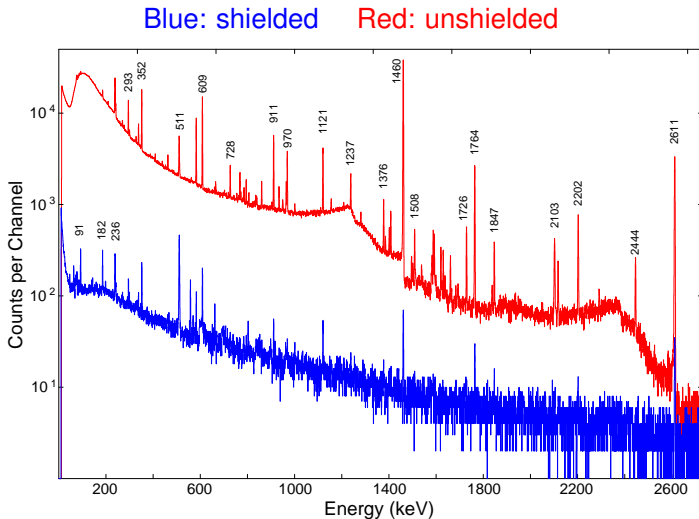
Blue:  $^{137}\text{Cs}$     Red:  $^{60}\text{Co}$



# GEARS: Germanium detector for Elemental Analysis and Radioactivity Studies



# Background suppression

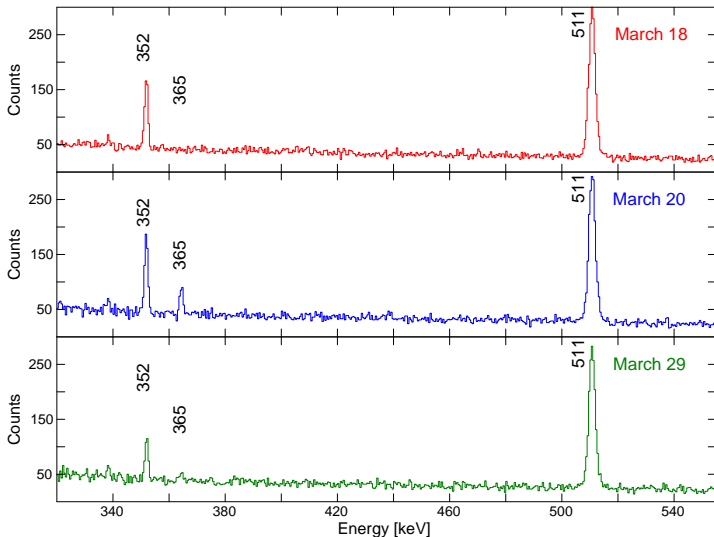


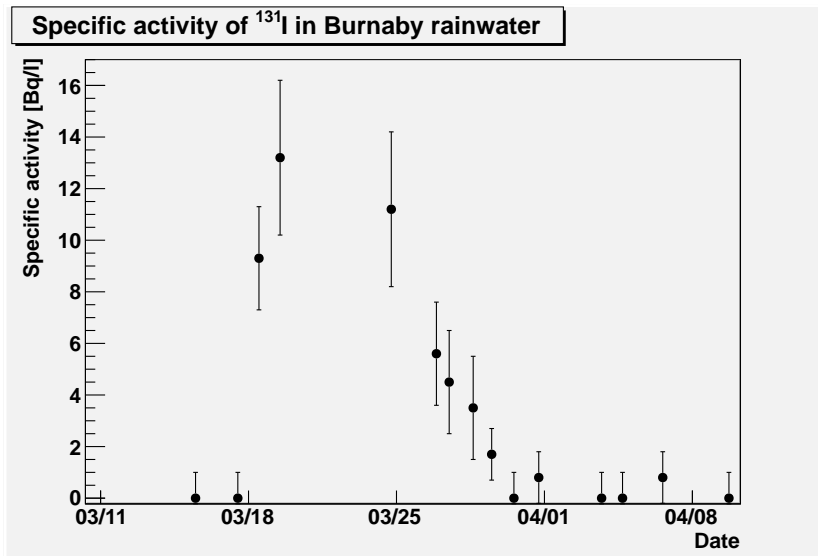


## Credits for sample collection and handling

- SFU Burnaby rainwater : SFU Nuclear Science graduate students
  - Aaron Chester
  - Rachel Ashley
  
- North Vancouver seaweed : SFU Physics
  - Ken Myrtle
  
- Bamfield seaweed : Bamfield Marine Sciences Centre
  - David Riddell
  - Hana Kucera

# $^{131}\text{I}$ signature in the SFU rainwater

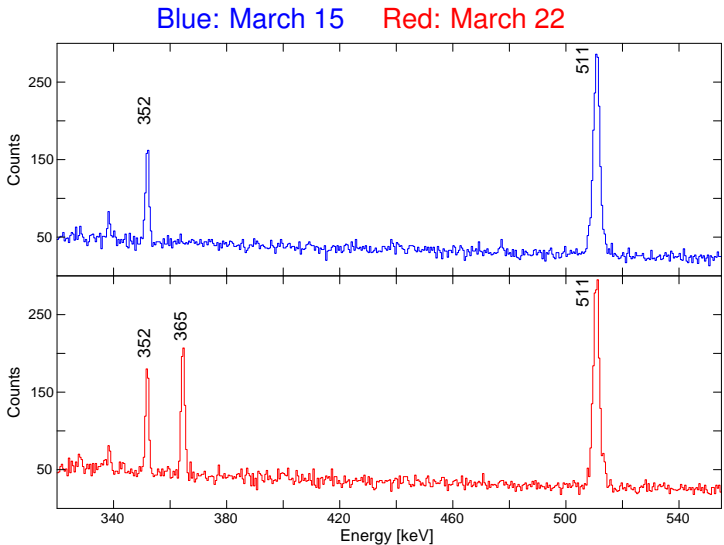


Time profile of the  $^{131}\text{I}$  signature in Burnaby rainwater

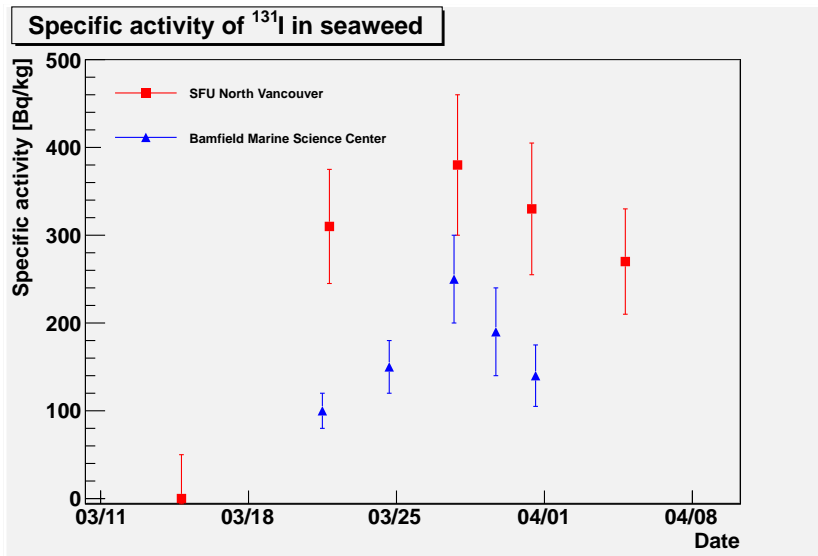
# $^{131}\text{I}$ sampling in seaweed



# $^{131}\text{I}$ signature in the North Vancouver seaweed



# Time profile of the $^{131}\text{I}$ signature in seaweed



# Time profile of the $^{131}\text{I}$ signature

